


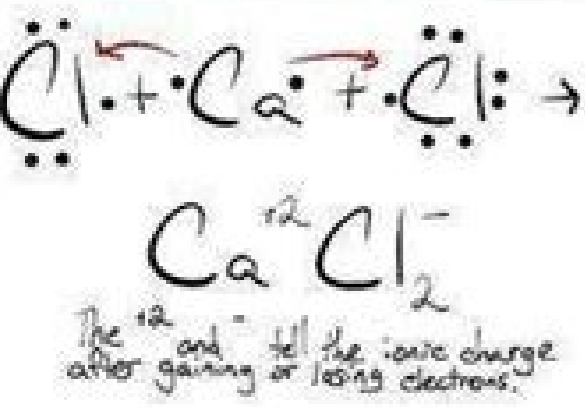
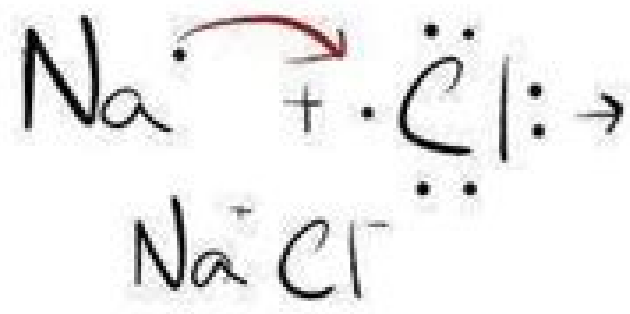
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Ionic VS Covalent

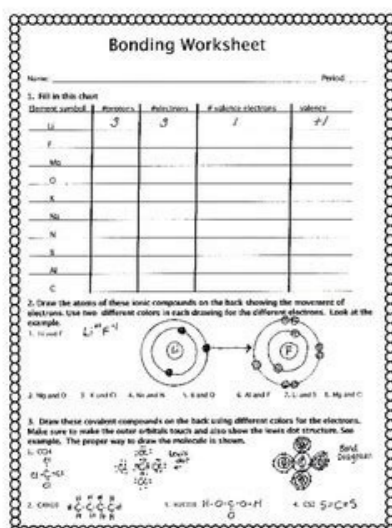
Metal + nonmetal → transfer of one or more e⁻ to have a stable octet of e⁻.

nonmetal + nonmetal → sharing e⁻ to have a stable octet of e⁻.



BONDS

@MrCoachK15 #KandZw



IONIC COMPOUNDS

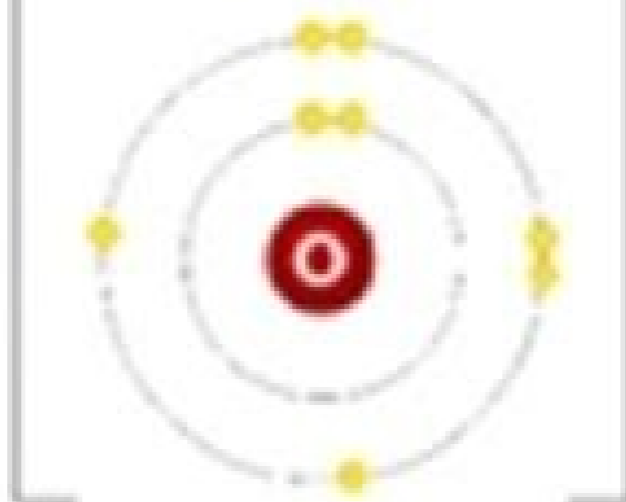
_____ can also be formed by _____ and _____

_____? The reason for this is so that they gain a full outer shell of electrons and become stable.

If we look at an oxygen atom for example, it has _____ in its valence shell. To become stable it could lose all six of these electrons or it could _____

And for _____? Which would be easier - gaining or losing electrons? _____

Which is easier?
- show what happens below.



When atoms _____ electrons they become _____ IONS are charged particles because they have a different number of _____ (+) and _____ (-).

_____ are atoms that have _____ or _____. Most atoms want to gain or lose electrons to become 'Happy' or _____

Atoms that _____ are called _____ and become _____ as they now have _____ electrons than protons.

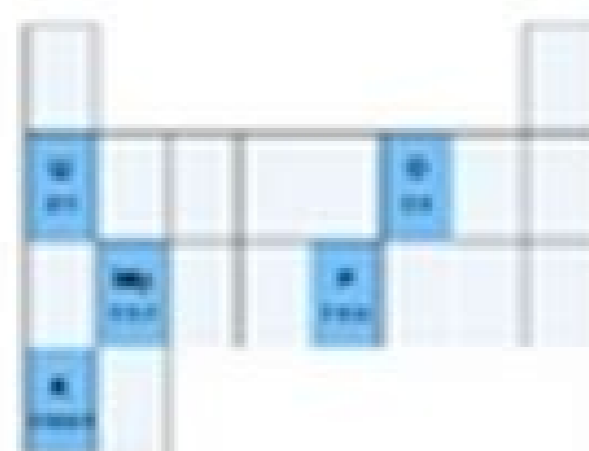
Atoms that _____ are called _____ and become _____ as they now have _____ electrons than protons.

Complete the table below

Atom							
	Na	Mg	Al	Si	P	S	Cl
What happens?	Loses 1 electron	Loses 2 electrons					
Ion	Na ⁺	Mg ²⁺					

_____ prefer to _____ and become _____ while _____ prefer to _____ and become _____

When _____ by _____ they obtain a noble gas configuration of electrons.



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